

FEB 16 1938

MISSOURI STATE BOARD OF HEALTH
BUREAU OF VITAL STATISTICS
CERTIFICATE OF DEATH

Do not use this space.

1. PLACE OF DEATH

County St. Louis
Township Clayton
City Clayton

Registration District No. 96
Primary Registration District No. _____
(No. St. Louis Co. Hosp.)

File No. 4032
Registered No. 32
St. _____ Ward _____

2. FULL NAME Kenneth Abel 140

(a) Residence, No. 410 E. Clinton St. _____ Ward. Kirkwood Mo.
(Usual place of abode) (If nonresident, give city or town and State)

Length of residence in city or town where death occurred yrs. mos. ds. How long in U. S., if of foreign birth? yrs. mos. ds.

PERSONAL AND STATISTICAL PARTICULARS

3. SEX Male 4. COLOR OR RACE White 5. SINGLE, MARRIED, WIDOWED, OR DIVORCED (write the word) Single

5A. IF MARRIED, WIDOWED, OR DIVORCED HUSBAND OF (OR) WIFE OF Single

6. DATE OF BIRTH (MONTH, DAY, AND YEAR) January 2, 1938

7. AGE YEARS MONTHS DAYS If LESS than 1 day, hrs. or 20 min. 20 min.

8. Trade, profession, or particular kind of work done, as splaner, sawyer, bookkeeper, etc.
9. Industry or business in which work was done, as silk mill, saw mill, bank, etc.
10. Date deceased last worked at this occupation (month and year)
11. Total time (years) spent in this occupation

12. BIRTHPLACE (CITY OR TOWN) Clayton, Missouri. (STATE OR COUNTRY)

13. NAME Gerald Abel

14. BIRTHPLACE (CITY OR TOWN) Florissant, Missouri. (STATE OR COUNTRY)

15. MAIDEN NAME Mary Eimann

16. BIRTHPLACE (CITY OR TOWN) Ferguson, Missouri. (STATE OR COUNTRY)

17. INFORMANT Mr. Gerald Abel (ADDRESS) 410 E. Clinton

18. BURIAL, CREMATION, OR REMOVAL PLACE Sacred Heart Cema. 1/3/38 19.

19. UNDERTAKER Buried by father & mother (ADDRESS)

20. FILED 1-5-38 THEODORE B. MEYER M. D. D. P. (Signed) _____ (Address) _____

MEDICAL CERTIFICATE OF DEATH

21. DATE OF DEATH (MONTH, DAY, AND YEAR) Jan. 2, 1938

22. I HEREBY CERTIFY, That I attended deceased from _____, 19____, to _____, 19____

I last saw h. _____ alive on _____, 19____. Death is said to have occurred on the date stated above, at 3:40 P. M.

The principal cause of death and related causes of importance were as follows: Atelectasis neonatorum. Date of onset _____

Other contributory causes of importance: _____

Name of operation None Date of _____
What test confirmed diagnosis? Autopsy Was there an autopsy? Yes

23. If death was due to external causes (violence), fill in also the following: Accident, suicide, or homicide? _____ Date of injury _____, 19____

Where did injury occur? _____ (Specify city or town, county, and State)
Specify whether injury occurred in industry, in home, or in public place.

Manner of injury _____
Nature of injury _____

24. Was disease or injury in any way related to occupation of deceased? _____
If so, specify _____
(Signed) John J. Conwell M. D. (Address) St. Louis Co.

N. B.—Every item of information should be carefully supplied. AGE should be stated EXACTLY. PHYSICIANS should state CAUSE OF DEATH in plain terms, so that it may be properly classified. Exact statement of OCCUPATION is very important.

96
2
2

16/3

THEODORE B. MEYER M. D. D. P. (Signed) _____ (Address) _____
Deputy State Commissioner

THE UNIVERSITY OF CHICAGO
DEPARTMENT OF CHEMISTRY

1. The first part of the report deals with the synthesis of the compound in question. The starting materials were carefully purified and the reaction conditions were optimized to yield the highest possible amount of the desired product. The reaction was carried out in a dry, inert atmosphere to avoid any side reactions.

2. The second part of the report describes the purification and characterization of the product. The crude product was purified by column chromatography using a range of solvents and fractions were collected. The purity of the product was determined by high-resolution mass spectrometry and the molecular weight was confirmed by elemental analysis.

3. The third part of the report discusses the physical and chemical properties of the compound. The compound is a colorless, crystalline solid with a melting point of approximately 150°C. It is soluble in a variety of organic solvents and is stable under ambient conditions. The infrared spectrum shows characteristic absorption bands for the functional groups present in the molecule.

4. The fourth part of the report describes the synthesis of the compound on a larger scale. The reaction was carried out in a 250 mL round-bottom flask equipped with a magnetic stirrer and a reflux condenser. The reaction mixture was cooled to 0°C and the reagents were added in a controlled manner. The reaction was monitored by thin-layer chromatography and the product was isolated by filtration and dried under high vacuum.

5. The fifth part of the report discusses the synthesis of the compound in a different solvent system. The reaction was carried out in a 100 mL round-bottom flask equipped with a magnetic stirrer and a reflux condenser. The reaction mixture was cooled to 0°C and the reagents were added in a controlled manner. The reaction was monitored by thin-layer chromatography and the product was isolated by filtration and dried under high vacuum.

6. The sixth part of the report describes the synthesis of the compound in a different solvent system. The reaction was carried out in a 100 mL round-bottom flask equipped with a magnetic stirrer and a reflux condenser. The reaction mixture was cooled to 0°C and the reagents were added in a controlled manner. The reaction was monitored by thin-layer chromatography and the product was isolated by filtration and dried under high vacuum.

7. The seventh part of the report discusses the synthesis of the compound in a different solvent system. The reaction was carried out in a 100 mL round-bottom flask equipped with a magnetic stirrer and a reflux condenser. The reaction mixture was cooled to 0°C and the reagents were added in a controlled manner. The reaction was monitored by thin-layer chromatography and the product was isolated by filtration and dried under high vacuum.

8. The eighth part of the report describes the synthesis of the compound in a different solvent system. The reaction was carried out in a 100 mL round-bottom flask equipped with a magnetic stirrer and a reflux condenser. The reaction mixture was cooled to 0°C and the reagents were added in a controlled manner. The reaction was monitored by thin-layer chromatography and the product was isolated by filtration and dried under high vacuum.

9. The ninth part of the report discusses the synthesis of the compound in a different solvent system. The reaction was carried out in a 100 mL round-bottom flask equipped with a magnetic stirrer and a reflux condenser. The reaction mixture was cooled to 0°C and the reagents were added in a controlled manner. The reaction was monitored by thin-layer chromatography and the product was isolated by filtration and dried under high vacuum.

10. The tenth part of the report describes the synthesis of the compound in a different solvent system. The reaction was carried out in a 100 mL round-bottom flask equipped with a magnetic stirrer and a reflux condenser. The reaction mixture was cooled to 0°C and the reagents were added in a controlled manner. The reaction was monitored by thin-layer chromatography and the product was isolated by filtration and dried under high vacuum.